

Advancements in Surface and Structural Engineering on Electrocatalysts for Enhanced CO₂ Reduction Reaction

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Abstract:

The electrochemical reduction of carbon dioxide (CO₂RR) presents a promising pathway for producing industrial chemicals using renewable electricity. Accelerating the deployment of CO₂RR technologies requires the development of highly selective, durable, and cost-effective catalysts. This study demonstrates that Cu-based catalysts modified with AuCu shell, or In₂O₃ interfaces exhibit significantly enhanced selectivity and stability for CO₂-to-CO conversion compared to pure Cu or reference catalysts. Au₂-Cu₈ alloyed core-shell catalyst (only 17% Au content) achieves a CO faradaic efficiency (FECO) of 94% at -0.8 V while Cu-In₂O₃ catalyst, in which a low amount of In₂O₃ decorated on Cu surface, achieves a FECO of 95% at -0.7 V. Theoretical calculations corroborate these findings, revealing that the CuAu and In₂O₃ components play crucial roles in lowering the energy barrier for the formation of key reaction intermediates. Collectively, our results highlight the potential of these engineered Cu-based catalysts for the efficient electrochemical conversion of CO₂ to CO, paving the way for sustainable industrial chemical production powered by renewable energy.

Keywords:

CO₂ reduction reaction, copper, indium oxide, gold, density functional theory.